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Total Synthesis of the Eight Diastereomers of the Syn-Anti-Syn Phytoprostanes F1 Types I and II

Siham El Fangour, Alexandre Guy, Valérie Despres, Jean-Pierre Vidal, Jean-Claude Rossi, and Thierry Durand*

UMR CNRS 5074, Universite´ *Montpellier I, Faculte*´ *de Pharmacie, 3e Etage-15, Avenue Charles Flahault, F-34093 Montpellier Cedex 05, France*

thierry.durand@univ-montp1.fr

Received November 7, 2003

Syntheses of the eight enantiomerically pure diastereomers of the syn-anti-syn phytoprostanes F_1 types I and II are described starting from D- and L-glucose. Key steps include Wittig coupling, Horner Wadsworth Emmons (HWE) reactions, and enantioselective reduction of α , β -unsaturated ketones.

Introduction

Isoprostanes (IsoPs) represent a new family of biomarkers for oxidative stress generated from peroxidation of polyunsaturated fatty acids via a free-radical-catalyzed mechanism.1,2 Higher plants generally do not synthesize the arachidonate precursor required for isoPs formation, but rather utilize α -linolenic acid for the formation of isoprostane F_2 -like compounds which have been termed phytoprostanes F_1 (PPF₁).³ Jasmonates are established plant signaling compounds inducing defense responses.⁴ Preliminary data indicate that phytoprostanes also induce phytoalexins in a variety of plant species, suggesting a possible function of phytoprostanes as mediators of plant defense reactions in response to oxidative stress.^{5,6} Since we were interested in assessing the physiological activities of each of the phytoprostanes F_1 types I and II, we found it more attractive to obtain sufficient quantities by chemical synthesis.

In 2003, we published a note describing the syntheses of *ent*-PPF1 type I **1** and its 16-epimer **2**. ⁷ We now report the syntheses of all eight diastereomers of the syn-antisyn PPF_1 types I and II starting from cyclopentane precursors **3** and **3**′, obtained from D- and L-glucose (Scheme 1).

Results and Discussion

The Phytoprostanes F_1 types I and II were identified from autoxidation of α -linolenic acid by Mueller.³ To confirm the stereochemistry of the eight enantiomerically pure diastereomers of phytoprostanes F_1 types I and II, and also to screen the physiological activity of these phytoprostanes, we have developed a general and flexible strategy from our common intermediate syn-anti-syn cyclopentane precursors **3** and **3**′ (Scheme 1).

Synthesis of *ent-***Phytoprostane F1 Type I 1 and 2 from D-Glucose.** The syntheses of *ent*-PPF₁ type I **1** and its 16(*S*) epimer **2** from the cyclopentane percursor **³** is shown in Schemes 2-4 and was published as a note.7 The first 9 steps leading to cyclopentane alkoxyester **3** were achieved in 27% overall yield by using our iodo pathway.8

The phosphorus synthon **7** was selected for the introduction of the upper chain of the PPF_1 type I, and was prepared by using the procedure outlined in Scheme 2.

The first step is the opening of the ϵ -caprolactone under acidic conditions leading to the corresponding hydroxy methyl ester **5**. Subsequent halogenation gave iodo ester **6**, which was transformed into the phophonium salt **7** in 98% yield in the presence of PPh₃ and a catalytic amount of K_2CO_3 .

The introduction of the upper chain (Scheme 3) was achieved by using the above phosphonium salt **7**, which

^{*} Address correspondence to this author. Phone: 33-4-67-54-86-23. Fax: 33-4-67-54-86-25.

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a Reagents and conditions: (a) MeOH, H₂SO₄, rt, 30 min, 94%. (b) I₂, PPh₃, imidazole, CH₂Cl₂ -10 °C, 2 h, 61%. (c) PPh₃, CH₃CN, 1% K_2CO_3 , 18 h, 98%.

reacts with the aldehyde **8** to afford the pure (Z)-enic ester **9** in 70% yield. No trace of trans compound could be detected by 13C and 1H NMR analyses.

The following reduction of the cis double bond of **9** under hydrogen atmosphere gave the diol **10**, which was protected in the presence of benzoyl chloride in pyridine in 77% yield.

The next steps were first a selective deprotection of the *tert*-butyldiphenylsilyl ether **11** with hydrogen chloride,¹¹ a Dess-Martin oxidation¹² of **12**, followed by a HWE coupling reaction in the presence of dimethyl-2 oxobutylphosphonate 14 ,¹⁰ yielding the trans- α , β -enone

ester **15** in 60% overall yield from the alcohol **12**. The relative configurations of the chiral centers were confirmed by homonuclear ¹H steady-state-difference NOE spectroscopy (DNOES) experiments (vide infra).

The diastereoselective reduction of the C16 keto group in **15** with the chiral reducing agents¹³ (*S*)- and (R) -BINAL-H gave the desired pure 16(*S*) derivative **16** and its 16(*R*) epimer **17** in 85% and 87% yield, respectively (Scheme 4).

Finally, saponification of ester functions, followed by treatment with an excess of $CH₂N₂$, afforded the desired *ent*-PPF₁ type I 1 and its 16 epimer 2 in 20% and 19% yields, respectively, from **3**, after 9 steps.

Synthesis of *ent***-Phytoprostanes** F_1 **Type II 31** and 32 from D-Glucose. Since the PPF₁ type II diaster-

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SCHEME 3. Synthesis of Precursor 15*^a*

a Reagents and conditions: (a) 1.1 equiv of DIBAL-H (1 M in toluene), toluene, -80 °C, 30 min, 97%. (b) 2.25 equiv of
thoxycarbonylpentyltriphenylphosphonium iodide 2.2 equiv of KN(SiMea)a THE -80 to 20 °C, 2 b, 70% (c) methoxycarbonylpentyltriphenylphosphonium iodide, 2.2 equiv of KN(SiMe3)2, THF, –80 to 20 °C, 2 h, 70%. (c) H2, Pd/C 10%, EtOH, 4
h, 98%. (d) 3 equiv of BzCl, pyridine, 20 °C, 1 h, 77%. (e) HCl 3% in MeOH, 20 °C, overnight equiv of dimethyl 2-oxobutylphosphonate 14, 3 equiv of NaN(SiMe₃₎₂, THF, 20 °C, 30 min, 60%.

SCHEME 4. Synthesis of *ent***-Phytoprostanes F1 Type I 1 and 2***^a*

^a Reagents and conditions: (a) (*S*)-BINAL-H, -100 °C, 2 h, 85%. (b) (*R*)-BINAL-H, -100 °C, 2 h, 87%. (c) 1 N NaOH, THF-MeOH, rt, 1 h, then CH2N2 91% for **1** and 93% for **2**.

a Reagents and conditions: (a) (i) *n*-BuLi, THF, -78 °C; (ii) diethylazelate, -78 °C, 3.5 h, 57%.

TABLE 1. $[\alpha]^{20}$ ^D $(1 \times 10^{-2}$, MeOH) for All Diastereomers **of PPF1 Types I and II**

diastereomer		α ²⁰ _D α ²⁰ _D	diastereomer
<i>ent</i> -PPF ₁ type I (1)			$-30 +25$ PPF ₁ type I (1')
ent-16-epi-PPF ₁ type I (2)	-13		+14 16- <i>epi</i> -PPF ₁ type I (2')
$ent-PPF1 Type II (31)$	-26		+23 PPF ₁ type II $(31')$
$ent-9\text{-}epi-PPF_1$ type II (32)	-7		$+7$ 9-epi-PPF ₁ type II (32')

eomers had never previously been prepared, we have performed the synthesis of these type II $PPF₁$ as shown in Schemes 6 and 7.

The *â*-ketophosphonate **19** was selected and prepared by using the following procedure published by Miftakhov et al*.* ¹⁴ and outlined in Scheme 5. The lithio derivative of dimethyl methylphosphonate **18** reacts with the diethylazelate to afford the corresponding *â*-ketophosphonate **19** in 57% yield.

The alkoxyester **3** was converted into the primary alcohol **20** by treatment with lithium aluminum hydride (LAH) in anhydrous ether in 94% yield (Scheme 6). The following protection, using toluenesulfonyl chloride, led to the corresponding tosylate **21** in 97% yield, which was reduced, once again, by LAH in refluxing ether to afford the methyl group of the α chain in 91% yield.

The introduction of the ω chain of the PPF₁ type II started by the regioselective deprotection of the bistriethylsilyloxy ether in the presence of *tert*-butyldiphenylsilyl ether, using ammonium fluoride in MeOH/THF to afford the diol **23** in 82% yield. The protection of the hydroxyl functions of **23** with benzoyl chloride in dry pyridine gave the diester **24** in 94% yield.

The *tert*-butyldiphenylsilyl ether **24** was converted into the alcohol **25** by treatment with a solution of 3% hydrogen chloride¹¹ in methanol/diethyl ether (1:1). Oxidation of the alcohol **²⁵** with the Dess-Martin periodinane12 provided the unstable aldehyde **26**, which was immediately used in the next step without purification.

The condensation of **26** with dimethyl 9-(ethoxycarbonyl)-2-oxononylphosphonate **19**, ¹⁴ in the presence of sodium hexamethyldisilyl amide, in anhydrous THF at room temperature, afforded the trans- α , β -enone ester **27** in 79% overall yield from alcohol **25** (Scheme 7). During the HWE reaction, it was not possible to avoid the formation of compound **28** derived from the elimination of the benzoyl group at C13, as already observed in other

^a Reagents and conditions: (a) 1.5 equiv of LAH (1 M in THF), diethyl ether, -78 °C, 1 h, 94%. (b) DMP, TEA, 1.5 equiv of TsCl, CH_2Cl_2 , rt, 2 h, 97%. (c) 1.1 equiv of LAH (1 M in THF), diethyl ether, reflux, 91%. (d) 4 equiv of NH4F, MeOH/THF (7:3), reflux, 2 h, 82%. (e) 3 equiv of BzCl, pyridine, rt, 1 h, 94%. (f) HCl 3% in MeOH/Et₂O (1:1), rt, overnight, 97%. (g) Periodinane, CH₂Cl₂, rt, 30 min. (h) 3 equiv of dimethyl 9-(ethoxycarbonyl)-2-oxononylphosphonate **19**, 2.8 equiv of NaN(SiMe₃)₂, THF, rt, 1 h, 79%.

series. The relative configurations of the chiral centers were confirmed by homonuclear 1H steady-state-difference NOE spectroscopy (DNOES) experiments (vide infra).

The diastereoselective reduction of the C9 keto group in **27** with the chiral reducing agents¹³ (R) - and (S) -BINAL-H gave the desired pure 9(*R*) derivative **29** and its 9(*S*) epimer **30** in 82% and 89% yield, respectively (Scheme 7).

Finally, deprotection of the benzoyl groups in **29** and **30**, in the presence of 1 N NaOH at room temperature, followed by excess of CH2N2, afforded the desired *ent*-PPF₁ type II **31** and its 9 epimer **32** in 38% and 42% yields, respectively, from the cyclopentane alkoxyester **3**, in 10 steps.

Syntheses of Phytoprostanes F1 Types I and II 1′**, 2**′**, 31**′**, and 32**′ **from L-Glucose.** All reactions of Dglucose have been duplicated with L-glucose leading to

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SCHEME 7. Syntheses of *ent***-Phytoprostanes F1 Type II 31 and 32***^a*

^a Reagents and conditions: (a) (*S*)-BINAL-H, -100 °C, 2 h, 89%. (b) (*R*)-BINAL-H, -100 °C, 2 h, 82%. (c) 1 N NaOH, THF/MeOH, rt, 1 h, then CH2N2 96% for **31** and 94% for **32**.

SCHEME 8. Syntheses of Phytoprostanes F1 Type I and Type II 1′**, 2**′**, 31**′**, and 32**′

the enantiomerically pure PPF_1 types I and II diastereomers **1**′, **2**′, **31**′, and **32**′ as shown in Scheme 8.

The physicochemical properties of the enantiomerically pure PPF1 types I and II diastereomers **1**′, **2**′, **31**′, and **32**′ were identical with those of **1**, **2**, **31**, and **32** except for the sign of specific optical rotation, as mentioned in Table 1.

Structural Determination of α , β -**Enones 15, 15'**, **27, and 27**′**.** The relative configurations of chiral centers were confirmed by homonuclear 1H steady-state-difference NOE spectroscopy (DNOES) experiments as shown in Figure 1.

For compound **15**, the relative cis configuration of the protons 11-H, 1-H, 4-H, and 6-H was determined by

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FIGURE 1. Observed NOEs resulting from irradiation of main protons are indicated with solid lines.

irradiation of 11-H, which induced NOEs of 1.3% on 1-H, 3.9% on 6-H, 5.2% on 4-H, and 0.7% on 7-H. Irradiation of 6-H induced NOEs of 1.0% on 1-H and 0.75% on 4-H. Similar results were observed for compound **15**′, the enantiomer of compound **15**.

Irradiation of 11-H induces a NOE of 6.3% on 6-H, 2.4% on 1-H, and 8.1% on 4-H, while irradiation of 6-H induces a NOE of 2.1% on 1-H and 1.3% on 4-H. These observations verify the relative cis configuration of the protons 1-H, 4-H, 6-H, and 11-H.

Concerning **27**, the irradiation of 6-H showed a NOE of 2.8% on 8-H, 2.1% on 4-H, and 0.88% on 1-H. Likewise, the irradiation of 8-H induces a NOE of 1.9% on 1-H and 1.3% on 4-H. These observations allow one to check the relative cis configuration of protons 1-H, 4-H, 6-H, and 8-H.

For compound **27**′, the enantiomer of compound **27**, the relative cis configuration of the protons 1-H, 4-H, 6-H, and 8-H was determined by irradiation of 6-H, which induced NOEs of 3.1% on 4-H, 1.8% on 1-H, and 2.4% on 8-H. Irradiation of 8-H induced NOEs of 3.0% on 1-H and 1.9% on 4-H.

Conclusion

In conclusion, the first synthesis of the eight enantiomerically pure diastereomers of the syn-anti-syn phytoprostanes F_1 types I and II has been accomplished starting form enantiopure alkoxyesters **3** and **3**′. Further studies of other phytoprostanes or analogues, as well as the assessment of individual biological activities in human and/or plant, are in progress and will be reported in due course.

Acknowledgment. We gratefully acknowledge the Ministère de l'Education Nationale et de la Recherche for financial support for one of us (S.E). The authors are grateful to Dr. A. Vidal for carefully reading the manuscript.

Supporting Information Available: Experimental procedures for syntheses and 1H NMR and 13C NMR data for compounds **⁵**-**7**, **⁸**-**17**, **¹⁹**, **²⁰**-**30**, and the target compounds **1**, **2, 31, 32, 1**′, **2**′, **31**′, and **32**′. This material is available free of charge via the Internet at http://pubs.acs.org.

JO035638I